

## Bridging to Polymers: Thermoset Lab Activity – Thermoset Lab Worksheet

### Activity Purpose

In this activity, we will make molecular “bridges” with different number of molecular connections and then we are going to explore their mechanical properties.

As shown in the table below, each group of 2-3 students is assigned particular ratios (the ratio terms are defined as EPOXY volume to AMINE volume, or Epoxy:Amine). By varying the ratios between the epoxy, our difunctional molecule, and amine, our multi-functional (>2) molecule, we can vary the mechanical properties.

After the ratios are mixed, they are poured into a mold, and the reactions are allowed to cure (or set) by being heating in a drying oven. Once the mold has completely reacted, its “stiffness” is tested using what is called a *flexural* test. We can relate how much weight is needed to flex the bar to the stiffness of the material.

### Group Ratios Table

Group	Epoxy Volume (ml)	Hardener Volume (ml)	Volume Ratio	Flexure mass Class Average (kg)	Flexure Weight Class Average (N)	Broke Bent Held 10kg
A1	10	0.5	20:1			
A2	10	2.5	4:1			
B1	10	1.0	10:1			
B2	10	3.0	3.33:1			
C1	10	1.5	6.66:1			
C2	10	3.5	2.86:1			
D1	10	2.0	5:1			
D2	10	5.0	2:1			

### Pre-Lab Questions

Before starting the activity, please complete the following questions using complete sentences.

- Why is it possible to vary the mechanical properties of our material?

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- Is the material that you are making a (thermoplastic or a thermoset)? *Circle the correct answer.*

3. In *your own words*, what is a thermoplastic?

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4. In *your own words*, what is a thermoset?

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5. In a laboratory, it is common to use a mass ratio instead of a volume ratio. If the density of epoxy and amine is 1.1 and 0.9 grams/mL, respectively, how much volume of each do you need to make a 10 gram sample with a 10:1 epoxy to amine mass ratio?

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6. What ratios were assigned to your group?

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7. To get stiff materials, what type of ratio(s) do you need (large or small or in between)?  
*Circle the answer(s) that apply.*

8. For flexible materials what type of ratio(s) do you need (small or large or in between)?  
*Circle the answer(s) that apply.*

### **Making Thermosets (sharing and interpreting data)**

Use the average data of other groups during all the periods testing during the day to fill out the final column of the data table.

On the graph on the next page, please plot the flexure weight (N) versus the amount of amine in the system (mL). Be sure to put units on axes (.5 increments), as well as label the axes.


**Post-Lab Questions**

Please complete the following questions using complete sentences.

1. Which ratio(s) of amine hardener is/are the stiffest? \_\_\_\_\_
2. Which ratio(s) of amine hardener is/are the softest? \_\_\_\_\_
3. What can you conclude about the amine hardener or what is the general trend between phr versus flexure weight (N)?

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4. Why does it follow that trend?

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5. Think back to the bridge analogy, why is there a difference in strength between ratios?

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